

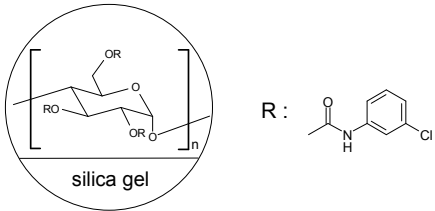


# INSTRUCTION MANUAL FOR CHIRALPAK® ID COLUMNS

Please read this instruction sheet completely before using this column

**ID columns can also be used in reversed phase mode.  
Please refer to the corresponding instruction sheet for details.**

## Column description

<b>CHIRALPAK® ID</b>
Amylose tris(3-chlorophenylcarbamate) immobilised on 5µm silica gel.

Shipping solvent: <b>n-Hexane / 2-Propanol solvent mixture (90:10 v/v)</b> All columns have been pre-tested before packaging. Test parameters and results, as well as the Column Lot Number, are included on a separate (enclosed) page.

**THIS INSTRUCTION SHEET IS NOT APPLICABLE TO ANY OTHER DAICEL COLUMNS**

## Operating Instructions

	<b>2.1x 150 mm 2.1x 250 mm</b> Analytical columns	<b>4.6 x 150 mm 4.6 x 250 mm</b> Analytical columns	<b>10 x 250 mm</b> Ⓢ Semi-prep. columns	<b>20 x 250 mm</b> Ⓢ Semi-prep. columns
Flow rate direction	As indicated on the column label			
Typical Flow rate	~ 0.1 - 0.2 ml/min	~ 1 ml/min	~ 5 ml/min	~ 18 ml/min
Pressure limitation	Should be maintained < 300 Bar (4350 psi) for maximum column life <a href="#">Adapt flow rates to column size.</a>			
Temperature	0 to 40°C			

## A - Mobile phases

CHIRALPAK® ID can be used *with all ranges of organic miscible solvents*, progressing from the traditional mobile phases used with other DAICEL columns (mixtures of alkanes/alcohol, pure alcohol or acetonitrile (CH<sub>3</sub>CN)) to mobile phases containing methyl *tert*-butyl ether (MtBE), tetrahydrofuran (THF), dichloromethane (DCM), chloroform (CHCl<sub>3</sub>), ethyl acetate (EtOAc) among others.

## B - Method Development - Screening

When developing methods we would recommend a screening approach.

1. The conditions described in Table 1 should be used as a **Primary Screening**.
2. If the compound or compound series are not soluble in any of these mobile phases, we recommend progressing directly to the **Secondary Screening** (Table 2).

**Table 1. Immobilised Primary Screening Solvents**

Primary solvent mixtures	Alkane <sup>①</sup> /2-PrOH	Alkane <sup>①</sup> /EtOH	Alkane <sup>①</sup> /MtBE/EtOH <sup>②</sup>	Alkane <sup>①</sup> /THF <sup>③</sup>	Alkane/DCM <sup>④</sup> /EtOH
Typical starting conditions	80:20	80:20	0:98:2	70:30	50:50:2
Advised optimisation range	99:1 to 50:50	99:1 to 50:50	80:20:0 to 0:40:60	95:5 to 0:100	85:15:0 to 0:80:20

① Alkane = n-Hexane, iso-Hexane or n-Heptane. Some small selectivity differences may sometimes be found.

② In absence of alkane, methanol is more efficient than ethanol when combined with MtBE.

③ In the case of no environmental restrictions, **use of DCM is preferred to THF** in terms of better enantioselectivity that the former may induce.

④ For excessively retained samples, addition of ethanol up to 20% in pure DCM would be helpful.

If a suitable chiral separation is not found using the **Immobilised Primary Screening** strategy, we recommend an **Immobilised Secondary Screening** to be applied using the following conditions:

**Table 2. Immobilised Secondary Screening Solvents**

Secondary solvent mixtures	EtOAc <sup>①</sup> /Alkane <sup>②</sup>	CH <sub>3</sub> CN <sup>③</sup> /Alcohol <sup>④</sup>
Typical starting conditions	50:50	100:0
Advised optimisation range	20:80 to 100:0	100:0 to 0:100

① Alcohols (④) or THF can be added into EtOAc to enhance the eluting strength for strongly retained compounds.

② Alkane: n-Hexane, iso-Hexane or n-Heptane. Some small selectivity differences may sometimes be found.

③ Transfers between alkane mixtures and CH<sub>3</sub>CN are preferably made with a transition in alcohol in order to avoid miscibility issues.

④ Alcohol: MeOH, EtOH and 2-PrOH.

**Note:** All solvent proportions indicated in this manual are by volume.

## C – General Comments

- ⇒ Additional solvent combinations such as CHCl<sub>3</sub>/Alkane, 1,4-Dioxane/Alkane, Toluene/Alkane or Acetone/Alkane can also be investigated with CHIRALPAK® ID column.
- ⇒ The typical starting conditions represent the mobile phases of upper middle eluting strength. Under such conditions, most of the analytes can be eluted within a reasonable time range with a good probability of full resolution of the enantiomers.
- ⇒ Toluene, MtBE and chlorinated solvents can be used in their pure form as the mobile phase.
- ⇒ For fast eluting solvents, such as THF, we recommend to add alkane in order to modulate the retention.
- ⇒ Detection with a regular UV detector may become difficult depending on a combination of sample and mobile phase (e.g. EtOAc, high percentages of DCM). In these cases an alternative detector, such as RI detector or ELSD (Evaporative Light Scattering Detector), may be more effective than the UV detector.

## D – Additives

For basic or acidic samples, it is necessary to incorporate an additive into the mobile phase in order to optimise the chiral separation.

❶ It has been found that certain amines, such as EDA and AE induce much better behaviour for certain basic compounds than the most commonly used DEA.

☞ The addition of a low percentage of an alcohol (e.g. 2% EtOH or MeOH) in the mobile phase may be helpful to ensure the miscibility of EDA and AE with the low polarity mobile phases.

Basic Samples require Basic additives	Acidic Samples require Acidic additives
Diethylamine (DEA) 2-Aminoethanol (AE)❶ Ethylenediamine (EDA)❶ Butyl amine (BA)	Trifluoroacetic acid (TFA) Acetic acid Formic acid
< 0.5% Typically 0.1%	< 0.5% Typically 0.1%

⇒ **STRONGLY BASIC solvent additives or sample solutions MUST BE AVOIDED**, because they are likely to damage the silica gel used in this column.

## Column care / Maintenance

- ❑ The use of a guard cartridge is highly recommended for maximum column life.
- ❑ Samples should be dissolved in the mobile phase. The mobile phase and the sample solution should be filtered through a membrane filter of approximately 0.5µm porosity to ensure that there is no precipitate before using.

### ☞ Column cleaning and regeneration procedures

Following extensive use of the column in multiple solvents there may be a change in separation reproducibility. In order to ensure consistent performance, a regeneration method may be implemented to eliminate any change in chiral recognition due to the history of the column (mobile phases, additives...).

- Flush with ethanol at 0.5 ml/min<sup>(\*)</sup> for 30 min, followed by 100% THF at 0.5 ml/min<sup>(\*)</sup> for 2 hours.
- Flush with ethanol at 0.5 ml/min<sup>(\*)</sup> for 30 min and then equilibrate with alkane/ethanol = 80/20 (v/v) prior to retesting the column.

If this is not successful, then try with 100% N,N-dimethylformamide (DMF) or N,N-dimethylacetamide (DMAC) at 0.3 ml/min<sup>(\*)</sup> for 3 hours instead of the THF flush.

<sup>(\*)</sup> Recommended flow rate for analytical columns (4.6mm ID).

## Column storage

- ❑ For column storage, remove the acidic or basic additives by flushing the column with the same mobile phase without the additive. Columns can be stored with additive-free mobile phases.

⇒ If you have any questions about the use of these columns, or encounter a problem, please contact [DAICEL CORPORATION](mailto:chiral@jp.daicel.com) for assistance ([chiral@jp.daicel.com](mailto:chiral@jp.daicel.com)).

**Operating this column in accordance with the guidelines outlined here will result in a long column life.**

**CHIRALCEL®**, **CHIRALPAK®** and **CROWNPAK®** are registered trademarks of **DAICEL CORPORATION**.

*Columns packed with 20µm material dedicated to preparative scale applications are also available from DAICEL CORPORATION.*

Column Name	Ref.	Product Type	Internal Diameter (mm)	Column Length (mm)
CHIRALPAK® ID 5µm	84311	Guard cartridge (x3)	4.0	10
	84324	Analytical	4.6	150
	84325	Analytical	4.6	250

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# INSTRUCTION MANUAL FOR CHIRALPAK® ID COLUMNS



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## <REVERSED PHASE>

**ID columns can also be used in normal phase mode.  
Please refer to the corresponding instruction sheet for details.**

### Switching between RP and NP mode

To switch from reversed phase mode to normal phase mode, and vice versa, column should be carefully flushed with miscible solvent.

It is highly recommended to apply the **regeneration procedure** described in the instruction sheet for normal phase mode. Before applying this protocol, any traces of salts should be removed by flushing with a mobile phase that does not contain any salts / buffers.

### Method Development / Reversed Phase

#### A - Mobile phases / For both UV and Mass detections

		ACIDIC (AMPHOTERIC) Compounds	NEUTRAL Compounds	BASIC Compounds ❶
CHIRALPAK® ID	Aqueous solution ❶	HCOOH aq. pH 2.0	Water	20mM NH <sub>4</sub> HCO <sub>3</sub> aq. pH 9.0 adjusted with a basic additive ❶
	Organic modifier ❷	CH <sub>3</sub> CN or MeOH or EtOH or IPA or THF		
	Typical starting conditions ❸	Aqueous solutions CH <sub>3</sub> CN	60%	40% ❹

☞ NOTE 1: If you cannot achieve sufficient resolution, try the complementary aqueous solutions

#### B – Complementary aqueous and buffer solutions / For UV detection

		ACIDIC (AMPHOTERIC) Compounds	NEUTRAL Compounds	BASIC Compounds ❶
CHIRALPAK® ID	Aqueous solution ❶	50mM Phosphate Buffer pH 2.0 OR H <sub>3</sub> PO <sub>4</sub> aq. pH 2.0 OR 100mM KPF <sub>6</sub> (or NaPF <sub>6</sub> ) aq. pH 2.0 adjusted with H <sub>3</sub> PO <sub>4</sub>	Water	20mM Borate Buffer pH 9.0 OR 20mM Phosphate Buffer pH 8.0 ❷ OR 100mM KPF <sub>6</sub> (or NaPF <sub>6</sub> ) aq.

☞ NOTE 2: The concentration of all the buffering salt should be less than 500mM.

- ❶ Refer to **section C** for preparation of aqueous solution and choice of basic additives.
- ❷  It is recommended to use CH<sub>3</sub>CN to start the investigation
- ❸  The elution power of organic modifiers for these columns is in the descending order of  
CH<sub>3</sub>CN > EtOH > MeOH: 50%CH<sub>3</sub>CN ≈ 65-70%EtOH ≈ 75-80%MeOH.

- ❑ The use of other organic solvents –**except THF**– has not been investigated and could be harmful to the columns.
- ❑ The use of alcohols causes the back pressure to be significantly higher compared to CH<sub>3</sub>CN due to their high viscosity in mixtures with water.
- ③ ❑ Retention can be adjusted by changing the proportion of CH<sub>3</sub>CN. Retention may be very sensitive to the amount of CH<sub>3</sub>CN present into the mobile phase.
  - ❑ Lowering the column temperature may increase the retention time and the selectivity.
  - ❑ Increasing the column temperature and decreasing the flow rate may increase the resolution.
- ④ ❑ **To maximize column life the use of a guard cartridge is essential when basic conditions are employed.**
  - ❑ The use of strong basic conditions (> pH 9) must be avoided, as they are known to damage the silica gel matrix.
  - ❑ When these columns are used at pH > 7, **the temperature should be maintained between 5°C and 25°C for maximum column life.**
- ⑤ High percentages of organic modifier in the mobile phase **may precipitate the buffering salt** from the solution, and lead to consequent clogging of the column (refer to the table below).

Water / Organic Modifier	Buffer solution / Organic Modifier
<b>90 / 10</b> to <b>0 / 100</b>	<b>90 / 10</b> to <b>15 / 85</b>

- ⑥ Do not use the phosphate buffer for pH > 8. When pH 9 is necessary, use the ammonium bicarbonate solution or borate buffer for maximum column life.

## C – Buffer preparation – Examples

### ➤ Preparation of pH 2 Phosphate buffer:

**Solution A:** 50mM potassium dihydrogenphosphate  
3.40g KH<sub>2</sub>PO<sub>4</sub> / FW 136.09, make up the volume to 500ml with HPLC grade water

**Solution B:** phosphoric acid (H<sub>3</sub>PO<sub>4</sub> 85% by weight)  
Adjust the pH of solution A to a value of 2.0 using solution B.

### ➤ Preparation of pH 2 KPF<sub>6</sub> (NaPF<sub>6</sub>) solution:

**Solution A:** 100mM potassium (sodium) hexafluorophosphate  
9.20g KPF<sub>6</sub> / FW 184.06 or 8.40g NaPF<sub>6</sub> / FW 167.95, make up the volume to 500ml with HPLC grade water

**Solution B:** phosphoric acid (H<sub>3</sub>PO<sub>4</sub> 85% by weight)  
Adjust the pH of solution A to a value of 2.0 using solution B.

### ➤ Preparation of pH 9 Ammonium bicarbonate solution:

**Solution A:** 20mM ammonium bicarbonate  
0.78g NH<sub>4</sub>HCO<sub>3</sub> / FW 78.05, make up the volume to 500ml with HPLC grade water

**Solution B** Basic additive such as diethylamine (DEA), triethylamine (TEA), ammonia (NH<sub>3</sub>) and so on.  
*\* DEA tends to give better peak shape than other bases.*

Adjust the pH of solution A to a value of 9.0 using solution B.

### ➤ Preparation of pH 8 Phosphate buffer:

**Solution A:** 20mM potassium hydrogenophosphate  
1.74g of K<sub>2</sub>HPO<sub>4</sub> / FW 174.18, make up the volume to 500ml with HPLC grade water

**Solution B:** 20mM potassium dihydrogenophosphate  
1.36g KH<sub>2</sub>PO<sub>4</sub> / FW 136.09, make up the volume to 500ml with HPLC grade water.

Adjust the pH of solution A to a value of 8.0 using solution B.

### ➤ Preparation of pH 9 Borate buffer:

**Solution A:** 20mM sodium tetraborate decahydrate  
3.81g of Na<sub>2</sub>B<sub>4</sub>O<sub>7</sub>·10H<sub>2</sub>O / FW 381.37, make up the volume to 500ml with HPLC grade water

**Solution B:** 20mM boric acid  
0.62g H<sub>3</sub>BO<sub>3</sub> / FW 61.83, make up the volume to 500ml with HPLC grade water

Adjust the pH of solution A to a value of 9.0 using solution B.

## Column care / Maintenance

- ❑ Any traces of salts should be removed before column storage and /or before switching to 100% organic solvent (use Water/CH<sub>3</sub>CN 60:40 (v/v) for instance)

**Refer main instruction for normal phase and column care/maintenance.**